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IN THE CLAIMS:

Please cancel Claim 31; amend Claims 2, 7, 12, 17, 25, and 30; and, add new dependent

Claims 61 - 65, as follows.

1. (Original) A method of preventing iridium particulates generated during etching of a layer

of iridium or an iridium compound in a plasma etch chamber from adversely affecting an etch

process subsequently performed in said plasma etch chamber, wherein said method comprises

exposing interior surfaces of said plasma etch chamber to a seasoning plasma generated from a gas

mixture comprising at least two gases selected from the group consisting of BCl₃, HBr, and CF₄.

2. (Currently Amended) The method of Claim 1, wherein said method further includes the step

of placing a dummy wafer having at least one iridium layer formed thereon in said plasma etch

chamber prior to prior to exposing surfaces of said chamber to said seasoning plasma.

3. (Original) The method of Claim 1, wherein said plasma etch chamber is a decoupled plasma

source etch chamber.

4. (Original) The method of Claim 3, wherein a plasma source power within the range of about

1000 W to about 1400 W and a substrate bias power within the range of about 150 W to about 250

W are applied during generation of said seasoning plasma.

5. (Original) The method of Claim 4, wherein said plasma source power and said substrate bias

power are applied for a time period within the range of about 30 seconds to about 120 seconds.

6. (Original) The method of Claim 1, wherein said gas mixture additionally includes argon.

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7. (Currently Amended) The method of Claim 6, wherein said gas mixture comprises BCl₃ is

provided at a flow rate of 30 seem, said HBr is provided at a flow rate of 30 seem, said CF4 is

provided at a flow rate of 30 seem, and said argon is provided at a flow rate of 40 seem, to said

plasma etch chamber.

8. (Original) The method of Claim 7, wherein a process chamber pressure within said plasma

etch chamber is maintained within the range of about 5 mTorr to about 10 mTorr.

9. (Original) The method of Claim 1, wherein said method further includes the step of cleaning

the plasma etch chamber with a purge gas prior to exposing surfaces of said chamber to said

seasoning plasma.

10. (Original) The method of Claim 1, wherein said method further includes the step of purging

said plasma etch chamber of remaining seasoning gas mixture after surfaces of said chamber to said

seasoning plasma.

11. (Original) A method of forming a storage capacitor in a plasma etch chamber, comprising

the steps of:

a) exposing interior surfaces of said plasma etch chamber to a seasoning plasma

generated from a gas mixture comprising at least two gases selected from the group consisting of

BCl₃, HBr, and CF₄;

b) purging said plasma etch chamber of remaining seasoning gas mixture;

c) loading a substrate having at least one iridium layer formed thereon into said plasma

etch chamber; and

d) plasma etching said at least one iridium layer.

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12. (Currently Amended) The method of Claim 11, wherein said method further includes the

step of placing a dummy wafer having at least one iridium layer formed thereon in said plasma etch

chamber prior to prior to exposing surfaces of said chamber to said seasoning plasma.

13. (Original) The method of Claim 11, wherein said plasma etch chamber is a decoupled

plasma source etch chamber.

14. (Original) The method of Claim 13, wherein a plasma source power within the range of

about 1000 W to about 1400 W and a substrate bias power within the range of about 150 W to about

250 W are applied during generation of said seasoning plasma.

15. (Original) The method of Claim 14, wherein said plasma source power and said substrate

bias power are applied for a time period within the range of about 30 seconds to about 120 seconds.

16. (Original) The method of Claim 11, wherein said gas mixture additionally includes argon.

17. (Currently Amended) The method of Claim 16, wherein said gas mixture comprises BCl₃

is provided at a flow rate of 30 seem, said HBr is provided at a flow rate of 30 seem, said CF₄ is

provided at a flow rate of 30 seem, and said argon is provided at a flow rate of 40 seem, to said

plasma etch chamber.

18. (Original) The method of Claim 17, wherein a process chamber pressure within said plasma

etch chamber is maintained within the range of about 5 mTorr to about 10 mTorr.

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19. (Original) A method of forming a storage capacitor in a plasma etch chamber, comprising

the steps of:

a) loading a substrate having at least one iridium layer formed thereon into said plasma

etch chamber;

b) plasma etching said at least one iridium layer;

c) removing said substrate from said plasma etch chamber;

d) cleaning said plasma etch chamber using a purge gas; and

e) exposing interior surfaces of said plasma etch chamber to a seasoning plasma

generated from a gas mixture comprising at least two gases selected from the group consisting of

BCl₃, HBr, and CF₄.

20. (Original) The method of Claim 19, wherein said method further includes the step of placing

a dummy wafer having at least one iridium layer formed thereon in said plasma etch chamber prior

to prior to exposing surfaces of said chamber to said seasoning plasma.

21. (Original) The method of Claim 20, wherein said plasma etch chamber is a decoupled

plasma source etch chamber.

22. (Original) The method of Claim 21, wherein a plasma source power within the range of

about 1000 W to about 1400 W and a substrate bias power within the range of about 150 W to about

250 W are applied during generation of said seasoning plasma.

23. (Original) The method of Claim 22, wherein said plasma source power and said substrate

bias power are applied for a time period within the range of about 30 seconds to about 120 seconds.

24. (Original) The method of Claim 19, wherein said gas mixture additionally includes argon.

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25. (Currently Amended) The method of Claim 24, wherein said gas mixture comprises BCl₃

is provided at a flow rate of 30 seem, said HBr is provided at a flow rate of 30 seem, said CF₄ is

provided at a flow rate of 30 seem, and said argon is provided at a flow rate of 40 seem, to said

plasma etch chamber.

26. (Original) The method of Claim 25, wherein a process chamber pressure within said plasma

etch chamber is maintained within the range of about 5 mTorr to about 10 mTorr.

27. (Original) A method of preventing platinum particulates generated during etching of a layer

of platinum in a plasma etch chamber from adversely affecting an etch process subsequently

performed in said plasma etch chamber, wherein said method comprises exposing interior surfaces

of said plasma etch chamber to a seasoning plasma generated from a gas mixture comprising at least

two gases selected from the group consisting of BCl₃, HBr, and CF₄.

28. (Original) The method of Claim 27, wherein said platinum particulates are generated during

etching of an electrode which includes platinum, iridium oxide (IrO₂), and iridium layers.

29. (Original) The method of Claim 27, wherein said gas mixture additionally includes argon.

30. (Currently Amended) A method of preventing etch byproduct particulates generated from

during etching a noble metal etch byproducts, which byproducts are nonvolatile at a substrate

temperature at which said noble metal is etched, and which accumulate in a plasma etch chamber

in which said noble metal is etched, from adversely affecting an etch process subsequently

performed in a said plasma etch chamber in which said metal was etched, wherein said method

comprises:

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a) placing a substrate in said plasma etch chamber , wherein said substrate specially

provides a source of a particulate entrapment or adhering material; and

b) exposing said substrate, chamber walls, and internal apparatus surfaces of said plasma

etch chamber to a seasoning plasma generated from a source gas that includes at least one principal

etchant gas used during an etch process which produced said nonvolatile etch byproducts, at a

substrate temperature that is equal to or greater than a substrate temperature at which said nonvolatile

etch byproducts were produced,

wherein whereby exposure of said substrate to said seasoning plasma generates an

entrapment and adhering material which adheres said nonvolatile etch byproducts to chamber walls

and internal apparatus surfaces.

31. (Cancelled)

32. (Original) The method of Claim 31, wherein said substrate includes an inorganic material

which provides a source for said entrapment and adhering material.

33. (Original) The method of Claim 32, wherein said method is performed at a substrate

temperature of 250°C or greater.

34. (Original) The method of Claim 32, wherein said method is performed at a substrate

temperature less than 250°C.

35. (Original) The method of Claim 31, wherein said substrate includes an organic material

which provides a source for said entrapment and adhering material, and wherein said method is

performed at a substrate temperature less than 250°C.

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36. (Original) The method of Claim 30, wherein said seasoning plasma includes a gas which

provides a source for said entrapment and adhering material.

37. (Original) The method of Claim 37, wherein said seasoning plasma includes a carbon-

containing gas.

38. (Original) A method of preventing particulates generated from metal etch byproducts, which

are nonvolatile at a substrate temperature at which said metal is etched, from adversely affecting an

etch process subsequently performed in a plasma etch chamber in which said metal was etched,

wherein said method comprises:

a) placing a substrate which includes a material which provides a source for an

entrapment and adhering material; and

b) exposing said substrate, chamber walls, and internal apparatus surfaces of said plasma

etch chamber to a seasoning plasma generated from a gas selected from the group consisting of Cl₂,

a chlorine-containing compound, and combinations thereof, at a substrate temperature that is equal

to or greater than a substrate temperature at which said nonvolatile etch byproducts were produced,

wherein exposure of said substrate to said seasoning plasma generates said entrapment and

adhering material which adheres said nonvolatile etch byproducts to chamber walls and internal

apparatus surfaces.

39. (Original) The method of Claim 38, wherein said substrate includes an inorganic material

which provides a source for said entrapment and adhering material.

40. (Original) The method of Claim 39, wherein said inorganic material is a dielectric selected

from the group consisting of silicon oxide, silicon nitride, aluminum oxide, and combinations

thereof.

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41. (Original) The method of Claim 39, wherein said method is performed at a substrate

temperature of 250°C or greater.

42. (Original) The method of Claim 41, wherein a metal which is etched within said plasma etch

chamber is selected from the group consisting of platinum, iridium, and combinations thereof, and

wherein said method is performed at a substrate temperature of at least 260°C, for a time period

ranging from about 2 minutes to about 30 minutes.

43. (Original) The method of Claim 39, wherein said method is performed at a substrate

temperature less than 250°C.

44. (Original) The method of Claim 43, wherein a metal which is etched in said plasma etch

chamber is copper, and wherein said method is performed at a substrate temperature of at least

210°C, for a time period ranging from about 2 minutes to about 30 minutes.

45. (Original) The method of Claim 43, wherein a metal which is etched in said plasma etch

chamber is selected from the group consisting of a nickel-iron alloy, a cobalt-iron alloy, and a nickel-

iron-cobalt alloy, and wherein said method is performed at a substrate temperature of at least 25°C,

for a time period ranging from about 2 minutes to about 30 minutes.

46. (Original) The method of Claim 38, wherein said substrate includes an organic material

which provides a source for said entrapment and adhering material, and wherein said method is

performed at a substrate temperature less than 250°C.

47. (Original) The method of Claim 46, wherein said organic material is a photoresist.

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48. (Original) A method of preventing particulates generated from metal etch byproducts, which

are nonvolatile at a substrate temperature at which said metal is etched, from adversely affecting an

etch process subsequently performed in a plasma etch chamber in which said metal was etched,

wherein said method comprises:

placing a substrate which provides a source of a dielectric material in said plasma etch

chamber; and

a)

b) exposing said substrate, chamber walls, and internal apparatus surfaces of said plasma

etch chamber to a seasoning plasma generated from a gas selected from the group consisting of Cl₂,

a chlorine-containing compound, and combinations thereof, at a substrate temperature that is equal

to or greater than a substrate temperature at which said nonvolatile etch byproducts were produced,

wherein exposure of said substrate to said seasoning plasma generates said dielectric material

which adheres said nonvolatile etch byproducts to chamber walls and internal apparatus surfaces.

49. (Original) The method of Claim 48, wherein said metal is selected from the group consisting

of platinum, iridium, copper, a nickel-iron alloy, a cobalt-iron alloy, a nickel-iron-cobalt alloy, and

combinations thereof.

50. (Original) The method of Claim 48, wherein said dielectric material is selected from the

group consisting of silicon oxide, silicon nitride, aluminum oxide, and combinations thereof.

51. (Original) The method of Claim 48, wherein said seasoning plasma further includes a noble

gas selected from the group consisting of argon, helium, xenon, krypton, and combinations thereof.

52. (Original) The method of Claim 51, wherein said seasoning plasma is generated from a gas

mixture comprising Cl₂ and argon, wherein Cl₂ comprises about 50 to about 90 volume %, and argon

comprises about 10 to about 50 volume %, of said gas mixture.

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53. (Original) The method of Claim 52, wherein Cl₂ comprises about 60 to about 80 volume %,

and argon comprises about 20 to about 40 volume %, of said gas mixture.

54. (Original) The method of Claim 51, wherein said seasoning plasma is generated from a gas

mixture comprising Cl₂, argon, and N₂, wherein Cl₂ comprises about 40 to about 90 volume %, argon

comprises about 10 to about 50 volume %, and N₂ comprises about 1 to about 20 volume %, of said

gas mixture.

55. (Original) The method of Claim 54, wherein Cl₂ comprises about 60 to about 80 volume %,

argon comprises about 10 to about 30 volume %, and N₂ comprises about 5 to about 20 volume %,

of said gas mixture.

56. (Original) The method of Claim 48, wherein said seasoning plasma is generated from a

source gas comprising Cl₂ and a chlorine-containing compound selected from the group consisting

of HCl, BCl₃, SiCl₄, and combinations thereof.

57. (Original) The method of Claim 56, wherein said source gas further includes a gas which

enhances dissociation of said chlorine-containing compound into active chlorine species.

58. (Original) The method of Claim 57, wherein said chlorine-dissociation-enhancing gas is

selected from the group consisting of N₂, NH₃, and combinations thereof.

59. (Original) The method of Claim 48, wherein said seasoning plasma is generated from a

source gas comprising a chlorine-containing compound selected from the group consisting of HCl,

BCl₃, SiCl₄, and combinations thereof, and wherein said source gas further includes a gas which

enhances dissociation of said chlorine-containing compound into active chlorine species.

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60. (Original) The method of Claim 59, wherein said chlorine-dissociation-enhancing gas is selected from the group consisting of N₂, NH₃, and combinations thereof.

- 61. (New) The method of Claim 1, wherein said gas mixture comprises BCl₃, HBr, and CF₄.
- 62. (New) The method of Claim 11, wherein said gas mixture comprises BCl₃, HBr, and CF₄.
- 63. (New) The method of Claim 19, wherein said gas mixture comprises BCl₃, HBr, and CF₄.
- 64. (New) The method of Claim 27, wherein said gas mixture comprises BCl₃, HBr, and CF₄.
- 65. (New) The method of Claim 29, wherein said gas mixture comprises BCl₃, HBr, CF₄, and argon.